Lithosphere versus asthenosphere mantle sources at the Big Pine Volcanic Field, California

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[1] Here we report the first measurements of the H₂O content of magmas and mantle xenoliths from the Big Pine Volcanic Field (BPVF), California, in order to constrain the melting process in the mantle, and the role of asthenospheric and lithospheric sources in this westernmost region of the Basin and Range Province, western USA. Melt inclusions trapped in primitive olivines (Fo₈₂–₉₀) record surprisingly high H₂O contents (1.5 to 3.0 wt.%), while lithospheric mantle xenoliths record low H₂O concentrations (whole rock <75 ppm). Estimates of the oxidation state of BPVF magmas, based on V partitioning in olivine, are also high (FMQ +1.0 to +1.5). Pressures and temperatures of equilibration of the BPVF melts indicate a shift over time, from higher melting temperatures (~1320°C) and pressures (~2 GPa) for magmas that are >500 ka, to cooler (~1220°C) and shallower melting (~1 GPa) conditions in younger magmas. The estimated depth of melting correlates strongly with some trace element ratios in the magmas (e.g., Ce/Pb, Ba/La), with deeper melts having values closer to upper mantle asthenosphere values, and shallower melts having values more typical of subduction zone magmas. This geochemical stratification is consistent with seismic observations of a shallow lithosphere-asthenosphere boundary (~55 km depth). Combined trace element and cryoscopic melting models yield self-consistent estimates for the degree of melting (~5%) and source H₂O concentration (~1000 ppm). We suggest two possible geodynamic models to explain small-scale convection necessary for magma generation. The first is related to the Isabella seismic anomaly, either a remnant of the Farallon Plate or foundered lithosphere. The second scenario is related to slow extension of the lithosphere.
1. Introduction

[3] The lithosphere is the strong lid at the surface of planet Earth [Barrell, 1914; Jordan, 1978]. This lid defines the different tectonic plates and consists of the crust and rigid uppermost mantle that moves on top of the viscous asthenospheric mantle. By definition these two mechanical layers are separated by some kind of rheological or thermal lithosphere-asthenosphere boundary (LAB) [e.g., Artemieva, 2006]. In most cases this boundary has been correlated with a seismic low velocity zone below [e.g., Rychert and Shearer, 2009; Rychert et al., 2010] that can be produced by high temperature [e.g., Ritzwoller et al., 2004], partial melt [Hammond and Humphreys, 2000; Jackson et al., 2004], and the presence of volatiles that reduce the velocity through anelastic effects [Karato, 2003; Hirschmann, 2010].

[3] To further understand the role of these different possibilities in the creation, evolution and seismic properties of the lithosphere-asthenosphere boundary (LAB), it is necessary to integrate seismological observations with petrological and geochemical interpretations. The Basin and Range Province in the western USA is a region where the lithosphere is actively evolving and where both lithospheric and asthenospheric mantle sources have been invoked to explain the geochemistry of widespread mafic volcanism in the last 10 Ma [e.g., Perry et al., 1987; Ormerod et al., 1988; Farmer et al., 1989; Fitten et al., 1991; Leeman and Harry, 1993; DePaolo and Daley, 2000]. This region has also recently been sampled by EarthScope’s Transportable Array, a leapfrogging array of broadband seismometers spaced approximately 70 km apart which allows for uniform and thorough imaging of upper mantle structure aiding in the investigation of the evolution of the LAB below continents and by temporary arrays focused on the Sierra Nevada.

[4] The tectonic processes and melting mechanisms that generated this mafic volcanism are still controversial. For example, mantle upwelling in response to plate extension is argued as one of the main processes for melt generation in the Basin and Range, similar to mid-ocean ridges [McKenzie and Bickle, 1988; Leeman and Harry, 1993; DePaolo and Daley, 2000; Wang et al., 2002]. This model is widely accepted because parts of the Basin and Range have been undergoing local episodes of extension for more than 30 Ma [e.g., Wernicke et al., 1987; McQuarrie and Oskin, 2010], and in several regions the timing of crustal extension coincides with changes in basalt composition that is consistent with a shift in the magma source from lithosphere to asthenosphere [DePaolo and Daley, 2000].

[5] However, other mechanisms have been suggested to explain the recent magmatism in the Basin and Range. Shear wave splitting data and topography have been interpreted as evidence for active mantle upwelling on a large scale [Savage and Sheehan, 2000; Lowry et al., 2000]. The passage of the Farallon slab may also drive asthenosphere upwelling [Ormerod et al., 1988]. A mantle melting profile across the Basin and Range, based on inverting basalt compositions [Wang et al., 2002], suggests mantle temperature variations as the primary control on the depth of melting, implying a dynamic convective process. Strong heterogeneities in the lithosphere [Farmer et al., 1989; Lee et al., 2001; Lee, 2005] may have accumulated throughout the Phanerozoic. Duerer et al. [2001] argued that volcanism in the Basin and Range occurs above Proterozoic sutures, which may contain easily fusible components. Dixon et al. [2004] and Humphreys et al. [2003] pointed to Farallon subduction as a major source of water to the lithosphere of the western USA, and as the ultimate control on volcanism, seismic structure and rheological
behavior. Many recent studies point to lithospheric foundering or drips [West et al., 2009; Levander et al., 2011; Crow et al., 2011], and that these “drips” have led to upwelling and thus magmatism, notably in southern California [Jones et al., 1994; Ducea and Saleeby, 1996; Zandt et al., 2004; Yang and Forsyth, 2006; Frassetto et al., 2011]. Also, the absolute motion of the North American Plate could produce small-scale convection that focuses melting and volcanism into SW-NE linear trends [Duerer et al., 2001]. Finally, numerical models suggest that mantle upwelling in high-shear regions like the western United States could be triggered by viscosity variations due to compositional heterogeneity, particularly mineral hydration in the mantle [Conrad et al., 2011].

[6] A petrological determination of the temperature and pressure conditions in the melting region is thus useful to understand the drivers of melting beneath the Basin and Range. These conditions are recorded in the major element composition (e.g., SiO2, FeO and MgO) of primitive mantle-derived magmas, but knowledge of Fe3+ and water contents of those magmas is necessary to provide accurate results. Melt inclusions, small (micro-meter sized) droplets of melt trapped during crystal growth, can be measured directly to obtain the Fe3+ and volatile (including water) contents of magma prior to ascent and degassing [e.g., Johnson et al., 2009; Kelley and Cottrell, 2009].

[7] The Big Pine Volcanic Field (BPVF) located in Owens Valley, California, east of the Sierra Nevada (Figure 1) is an ideal study site for determining melting conditions in the western Basin and Range for several reasons. (1) The BPVF consists of over 20 young volcanic vents that range from ∼1.3 Ma to 32 ka and produced basaltic lavas and tephras with abundant mantle xenoliths [Ormerod et al., 1991; Beard and Glazner, 1995; Blondes et al., 2007; Kirby et al., 2008]. This is ideal for obtaining undegassed melt inclusion because mantle xenolith transport and pyroclastic deposits require rapid magma ascent, which limits H2O-loss from the magma and melt inclusion. (2) The region also has been affected by Farallon subduction (18–20 Ma), Basin and Range extension beginning at ∼12 Ma, and transtension ongoing since the Pliocene [Stockli et al., 2003; Phillips and Majkowski, 2011; Jones et al., 2004], providing information about the relationship of magmatism to tectonism. (3) Blondes et al. [2008] concluded that crustal contamination is not a major process for the BPVF, based on radiogenic isotopes, trace element systematics, uniform MgO contents (>6 wt%) and the abundance of mantle xenoliths. (4) Several vents in the BPVF display monotonic and large temporal variations in magma chemical composition, consistent with binary mixing of distinct mantle melts over the course of single eruptive episodes [Blondes et al., 2008].

[8] Many lines of evidence point to lithospheric thinning beneath the BPVF. Shallow mantle here has low P and S velocities [e.g., Wernicke et al., 1996; Jones and Phinney, 1998; Savage et al., 2003; Boyd et al., 2004; Yang and Forsyth, 2006]. If this anomaly reflects hot or partially molten asthenosphere, little room is left for a cold and rigid mantle lithosphere below the Moho. Ormerod et al. [1991] used trace element inverse modeling to exclude garnet in the source, limiting melting depth to <90 km. Spinel-bearing peridotite xenoliths hosted in the BPVF lavas record Ca-in orthopyroxene temperatures of 1000–1100°C, and equilibrium depths as shallow as 35 km, indicating a lithospheric origin [Lee et al., 2000]. Some of the lavas contain clinopyroxenes that last equilibrated with melt as deep as ∼55 km [Mordick and Glazner, 2006] suggesting rapid melt extraction and cooling above this region. This suite of observations point to the Big Pine volcanics as recorders of a thinned lithosphere-asthenosphere (LAB) boundary.

[8] Here, we present quantitative petrologic information on primitive samples from the BPVF, guided by our new data on melt inclusions hosted in olivine (Fo 82–90) and water contents in mantle xenoliths. These new results are combined with surface wave inversions that constrain the mantle shear velocity structure. We discuss possible melting mechanisms within the context of evolving lithosphere and asthenosphere mantle in the recent volcanism in the Basin & Range by exploring these fundamental questions at BPVF: (1) Does the melt composition change at BPVF with depth, reflecting distinct chemical boundaries in the mantle? (2) What is the relationship between petrologically derived T and P estimates and the rheological changes at the LAB? (3) Is the LAB a boundary where mantle rheology changes, and/or an area where melt segregates and accumulates?

2. Samples, Data and Analytical Methods

[10] Scoria samples were collected from cinder cones from the BPVF in 2008 (locations shown in
Because of the focus on volatile concentrations, fine lapilli and ash deposits were targeted, as melt inclusions from these samples ascended and cooled rapidly (minutes to hours), thus limiting the time available for diffusive loss of H$_2$O through olivine. All melt inclusions were selected from pyroclasts <2 cm in diameter, which based on our recent finding [Lloyd et al., 2010], experience minimal H$_2$O loss during ascent and cooling.

[11] After crushing and sieving, olivines were hand-picked under a binocular microscope to select those with naturally glassy, fully enclosed, >30 micron melt inclusions. Inclusions were exposed, individually polished, and mounted in indium for volatile analysis by ion probe at the Carnegie Institution of
Washington, major elements by electron probe at the American Museum of Natural History (AMNH), and trace elements by laser ablation ICP-MS at LDEO, in that order (see auxiliary material for data and analytical conditions). Host olivines, augmented by others from the same sample to expand the population, were also analyzed for major and trace elements and volatile contents (data in the auxiliary material).

3. Data Treatment

3.1. Oxidation State

Constraining magma oxidation state is crucial not only in evaluating melt inclusion and host olivine equilibrium and correcting for post-entrapment crystallization within the inclusion, but also in calculating primary compositions in equilibrium with the melting mantle. Critical to all of these issues is the proportion of Fe that exists as Fe$^{3+}$ vs Fe$^{2+}$, as only Fe$^{2+}$ strongly affects olivine-melt equilibrium. Direct measurement of the Fe$^{3+}$/Fe$^{T}$ (total) ratio in melt inclusions is now possible through micro-Xanes [Kelley and Cottrell, 2009], although this technique is still not widely available for large throughput measurements. Thus, we apply here another oxybarometer, the partition coefficient of V between melt and olivine [e.g., Canil, 2002; Mallmann and O’Neill, 2009]. Vanadium may exist as multiple species in magmas, V$^{2+}$, V$^{3+}$, V$^{4+}$ and V$^{5+}$, the proportions of which depend on fO$_2$. In olivine, compatibility increases from V$^{5+}$ to V$^{4+}$ to V$^{3+}$, and from this, the bulk V partitioned between co-existing olivine and melt has been calibrated as a function of fO$_2$. We use the expression in Canil [2002], which agrees well with the parameterizations in Mallmann and O’Neill [2009] for a wide range of fO$_2$ and compositions.

Figure 2 shows the results of the D$_{(V)ol/liq}$ oxybarometer. For three cinder cones with the least evolved compositions (MgO > 6% wt: Jalopy, Scarlett and Quarry cones), olivines and melt inclusions were analyzed for V by laser ablation ICP-MS (Table S7 in in the auxiliary material). In Jalopy and Quarry cones, V in olivine increases slightly as forsterite content decreases, but concentrations are uniformly lower in Jalopy than Quarry despite similar V contents in the melt inclusions (Figure 2a). This leads to a lower D$_{(V)}$, and thus higher fO$_2$ (FMQ + 1.6, Figure 2b), for Jalopy cone than for Quarry cone (FMQ + 0.9, Figure 2b). In Scarlett cone olivines, D$_{(V)}$ and fO$_2$ are intermediate between the other two (Figure 2b). The oxidation states estimated from V predict significant Fe$^{3+}$ in these magmas, from 23–30% Fe$^{3+}$/Fe$^{T}$ [Kelley and...]

Pyroxenes from lithospheric mantle xenoliths from the well characterized Oak Creek location (reported in Lee [2005]) were also analyzed for major and trace elements and volatile contents (data in the auxiliary material).
Independent support for the high oxidation state of Jalopy magmas comes from the very high sulfur contents of the least degassed melt inclusions ([C24]500 ppm S; Table S1), consistent with FMQ > +1 [Jugo et al., 2010]. In their most recent work, Kelley and Cottrell [2012] reported four micro-Xanes analyses of Jalopy melt inclusions (different inclusions from the same sample reported here), and obtain Fe³⁺/Fe₀ of 24–25%. These data, along with a broader survey of coupled V-Fe³⁺ measurements in Kelley and Cottrell [2012], support the use of the V oxybarometer, and the oxidized nature of Big Pine magmas.

3.2. Equilibrium and Post-entrapment Crystallization

It is important to evaluate the equilibrium of a melt inclusion with its host olivine because it is possible that olivine may precipitate at/or near the host walls or elements may exchange with the olivine and melt during cooling [e.g., Danyushevsky et al., 2002]. For this, we first evaluate the Fe-Mg equilibrium with the host olivine by comparing the Mg# (before any correction) of the melt inclusion versus the Fo content of the olivine host. Melts in equilibrium with the host olivine should lie on the equilibrium line in Figure 3 (using a constant K_D^ol-lig (Fe²⁺/Mg) = 0.3 and the Fe³⁺/Fe₀ = 20–30% for each melt inclusion as determined from the V proxy. The rest of the melt inclusions require post entrapment crystallization correction on average of <6%.

Notably, most melt inclusions from Jalopy and Scarlett cones erupted in equilibrium with their host olivines, even for very primitive compositions (Fo₈₉–₉₀). This is consistent with the fast cooling of the small fraction collected (ash and <2 cm lapilli), and rapid ascent in the volcanic conduit. On the other hand, all the inclusions from Fish Springs cone reflect post entrapment crystallization, requiring a correction on average of <6% (Figure 3). These melt inclusions were corrected for post-entrapment crystallization by adding olivine until equilibrium with the host olivine was reached. The corrected values are reported in the auxiliary material (Table S1).

4. Results

4.1. Major and Trace Element Composition and Fractional Crystallization

After post-entrapment crystallization correction, the olivine hosted melt-inclusion data range from basalts and trachy-basalts to basaltic trachy-andesites, similar to the bulk rocks of the BPVF (Figure 4a). Almost all of the BPVF volcanics and melt inclusions belong to the alkaline series (high K₂O > 1 wt% and nepheline normative). The melt inclusion data are in the same range as the bulk rock chemistry from the same vent with the exception of Jalopy cone, where melt inclusions have uniformly higher alkalies and lower SiO₂ and MgO (Figure 4a). The samples with MgO values >6% are consistent with modeled liquid lines of descent (LLD) controlled mostly by olivine crystallization (Figures 4b–4d). The systematic decrease in FeO and CaO, and increase in Al₂O₃ in samples with MgO < 6% is consistent with clinopyroxene and magnetite fractionation (Figures 4b and 4c). These systematics are expected given the sample petrography, where olivine is the main crystalline phase in BPVF lavas, followed by minor (<5%) clinopyroxene and plagioclase in some of the more fractionated samples [Blondes et al., 2008; this study]. Fish Springs cinders also contain amphibole crystals. Also, within the Jalopy melt inclusions there is a minor population with high CaO (Figure 4b). Similar high-CaO inclusions have been found worldwide, typically in Fo > 86 olivines, often with anomalous major and trace elements [Schiano et al., 2000; Danyushevsky et al., 2002]. These high-Ca inclusions, along with those that bear clear evidence of clinopyroxene fractionation, were excluded from the petrologic modeling below because their compositions are too evolved from primitive mantle.
Figure 4. Major element composition (after post-entrapment crystallization correction) of the BPVF melt inclusions compared with bulk rock data from Blondes et al. [2008] and this study (Table S1). The samples with MgO values >6% are consistent with modeled liquid lines of descent (LLD) controlled by olivine crystallization, while more evolved samples show evidence for clinopyroxene and magnetite fractionation (cpx + mt vectors in Figures 4b–4d). The liquid lines of descent were modeled using Petrolog Software [Danyushevsky and Plechov, 2011], with the parameters of Beattie [1993]. In Figure 4b notice that there are two main populations of samples; the vents older than 0.7 Ma are on olivine control trends with higher FeO values at a given MgO content than the samples younger than 0.5 Ma. The gray area indicates possible primary melt compositions in equilibrium with a mantle source with Fo90 assuming Fe3+/FeT of 10–30%. Considering the highly oxidized nature of the BPVF (Fe3+/FeT = 20–30%) some of the primitive samples (MgO > 7%) are nearly primary in composition. Also notice in Figure 4d the minor population of high CaO melt inclusions within Jalopy vent data.
melts (see Lee et al. [2009] for more details on sample selection).

There are two main populations of samples at the BPVF, those from vents older than 0.5 Ma (e.g., Papoose, Oak Creek A and C) and those younger (including all the cones studied here that yielded melt inclusions). The older suites lie on olivine control trends with higher FeO values at a given MgO content than the younger samples (Figure 4b). Considering the highly oxidized nature of the BPVF (Fe$^{3+}$/Fe$^{2+} = 20–30\%$) some of the more magnesian samples (MgO > 7\%) are nearly primary in composition (gray area in Figure 4b), making this suite of samples ideal to evaluate melting processes that produced the BPVF.

The melt inclusion data from Fish Springs, Red Mountain and Prairie have the same trace element patterns as the host tephra (Figures 5a–5d and Table S4 in the auxiliary material). The Scarlet melt inclusions have the same overall patterns but they are generally more enriched than the bulk-rock data from a lava flow from the same vent (Figure 5d). This relative enrichment is consistent with fractional crystallization because the lava has MgO close to 8 wt\% and the melt inclusions and host tephra have less than 7\% (Figure 4). Although we do not have data for the exact host tephra that yielded the Quarry melt inclusion (Figure 5e), the overall enrichments compared to bulk-rock data from the same deposit [Blondes et al., 2008] are also consistent with fractional crystallization (see Figure 4). The Jalopy melt inclusion data show much more diverse trace element compositions compared with both the host tephra and data from the same vent reported in the literature (Figure 5f). This is consistent with the differing major element compositions as well. All of the BPVF melt inclusions show the relative enrichments of fluid mobile elements (e.g., Pb, Ba) and depletions in high-field strength elements (e.g., Nb, Ta) typical of subduction zone magmas (Figure 5).

4.2. Volatile Content and Degassing

Although melt inclusions are trapped inside olivine hosts, they can be closed at various stages of degassing as magma ascends to the surface. The different solubility of the volatile species allows a quantitative evaluation of melt inclusion degassing [e.g., Dixon and Stolper, 1995]. CO$_2$ is less soluble than H$_2$O and initial ascent and vapor exsolution leads to rapid loss of CO$_2$ from the melt, with little change in H$_2$O. This kind of degassing path is observed for the Jalopy melt inclusions (Figure 6a), and is quantitatively consistent with closed-system degassing from >5 kb (~20 km) to the surface (using the VolatileCalc model in Newman and Lowenstern [2002]). The H$_2$O-S data also define a very coherent degassing trend for Jalopy melt inclusions, with coupled loss of H$_2$O and S (Figure 6b). Given the very high CO$_2$ and S contents (Figures 6a and 6b), the good fit to the degassing path, and the high Fo contents of some of the olivine hosts (~Fo$_{90}$), there is a high degree of confidence in the initial water content of the Jalopy magma, at ~2.2 wt\% H$_2$O. Such water contents are surprisingly high for continental basalts, overlapping the low end of the range for arc basalts and the upper end for back-arc basin basalts [Zimmer et al., 2010]. On the other hand, Scarlett cone inclusions appear to lie on a separate degassing path, also starting at ~5 kb and with ~3000 ppm CO$_2$, but at lower water contents, ~1.5 wt\% H$_2$O (Figure 5b).

Additional support for this kind of primary variation in volatile concentrations comes from the H$_2$O-Cl systematics (Figure 6c). Cl has a higher solubility in vapor-saturated basaltic melts than H$_2$O, and BPVF melt inclusions form linear arrays reflecting H$_2$O loss (through either degassing or diffusion) at nearly constant Cl. The melt inclusions with the maximum H$_2$O contents define a common H$_2$O/Cl ratio of 66 for the BPVF, which is the upper bound of H$_2$O/Cl observed in arc and OIB melt inclusions [Ruscitto et al., 2012; Stolper et al., 2004; Koleszar et al., 2009]. The common H$_2$O/Cl ratio provides a useful baseline with which to estimate initial water contents even for partially degassed inclusions. By this argument, Fish Springs cone magmas would have the highest H$_2$O contents at Big Pine, on the order of 3 wt\% H$_2$O. This cone also erupts among the most evolved magma and so these higher water contents may reflect some crystal fractionation. Taken together, the Big Pine melt inclusions record a range of undegassed H$_2$O contents (from 1.5 to 3 wt\%) that vary systematically from cone to cone and are surprisingly high given the fact that BPVF is not in a subduction setting today.

In order to characterize the water contents of the lithospheric mantle below BPVF we measured water concentrations in pyroxenes from well characterized mantle xenoliths hosted in the 1.3 Ma Oak Creek flow [Lee et al., 2000; Lee, 2005]. The average cpx and opx H$_2$O contents are below 250 ppm and 150 ppm, respectively, defining an opx/cpx ratio of 0.4 (similar to the value of 0.5 measured experimentally by Aubaud et al. [2004] and recalibrated by Aubaud et al. [2007], but lower than
the 0.5–0.9 range measured experimentally by Tenner et al. (2009)). Plotted individually there is a strong negative correlation between Mg# and water contents in the cpx, consistent with a melt removal process that depletes the solid in water (an incompatible element) as Mg# increases (Figure 7a).

Reconstructed bulk peridotite H₂O concentrations in mantle xenoliths from the available four locations in the Western USA (Dish Hill, Grand Canyon, San Carlos from Li et al. [2008] and BPVF from this study) are plotted in Figure 7b. Only the Grand Canyon samples have water contents as high as...
Figure 6. Volatile contents and degassing path for BPVF melt inclusions. (a and b) The degassing path for the Jalopy melt inclusions consistent with closed-system degassing from >5 kb (~20 km). The very high CO₂ and S contents together with the good fit to the degassing path, and the high Fo contents of some of the olivine hosts (~Fo89) suggest that initial water content of the Jalopy magma is ~2.2 wt% H₂O. (c) The melt inclusions with the maximum H₂O contents at each vent define a common H₂O/Cl ratio of 66 for the BPVF and provide a useful baseline to estimate initial water contents for even partially degassed inclusions. Bold symbols are candidates for undegassed H₂O.

Figure 7. Water concentration in (a) mantle pyroxenes and (b) bulk peridotites from Big Pine (this work) and available data from the Western USA [Li et al., 2008]. (a) Comparison of H₂O concentrations in cpx versus opx, where Big Pine, Dish Hill and San Carlos fall along opx/cpx H₂O = 0.4 line, and Grand Canyon pyroxenes plot at higher opx/cpx. Big Pine points are averages ±1 σ of several measurements within one pyroxene from three different samples. The inserted figure on the bottom right corner illustrates the strong co-variation in H₂O versus Mg# for individual cpx data points from Big Pine. (b) Bulk peridotite H₂O concentrations in mantle xenoliths from four locations in the Western USA, calculated from modes and mineral concentrations (from Li et al. [2008]; and for Big Pine: modes of 53% ol, 28% cpx, 15% cpx and 2% spinel and H₂O concentrations in pyroxenes as measured and in olivine assuming ol/cpx H₂O = 0.036 as in Li et al. [2008]). Big Pine mantle xenoliths have uniformly lower H₂O concentrations than depleted MORB mantle (DMM), compiled from Saal et al. [2002], Simons et al. [2002], Salters and Stracke [2004], and Workman and Hart [2005].
4.3. Source Lithology

[24] Determining the magma source lithology is crucial to petrological modeling based on parameterization of experimental data and observations made on mantle peridotite sources. However, exposed sections of the upper mantle often contain pyroxenites, which can dramatically affect the melt productivity and major element composition of the resulting melts [Schulze, 1989; Hirschmann and Stolper, 1996; Kogiso et al., 2004]. Isotopic studies of global basalts also suggest the presence of a recycled or metasomatic component other than peridotite in the convecting mantle [Hofmann and White, 1982; Hauri, 1996; Chauvel et al., 2008; Jackson and Dasgupta, 2008].

[25] In recent years, the development of new petrological methods based on major element geochemistry of melts and trace element compositions of olivine [e.g., Sobolev et al., 2005; Dasgupta et al., 2007; Herzberg and Asimow, 2008] have provided new information on the composition of mantle source lithology. Sobolev et al. [2005, 2007] suggested that an olivine-free pyroxenite (eclogite) source can be identified from the composition of melts that crystallized olivines having higher Ni, lower Ca, and high Fe/Mn relative to peridotite-derived melts.

[26] Herzberg [2011] produced a new complete parameterization to discriminate source lithology based on olivine compositions built on the Sobolev et al. [2005, 2007] analytical work. According to these parameterizations, most of the BPVF olivines analyzed in this study have Ni, Mn contents and Fe/Mn consistent with magmas produced from a peridotite source (Figure 8). The relatively elevated Ni, Fe/Mn and low Mn contents from olivines from the Fish Springs, Prairie, Red Mountain vents and a minor subset of Jalopy vent suggest significant fractionation or even a possible pyroxenite component. The samples suspected to have a pyroxenite source were excluded from the thermodynamic modeling below because it was calibrated on experimental work on mantle peridotite – melt equilibrium [Lee et al., 2009].

[27] Zn/Fe has also been suggested to be a useful discriminant of peridotitic versus pyroxenitic sources [Le Roux et al., 2010]. Whole-rock Zn/Fe systematics of BPVF primitive basalts are also consistent with a peridotite source [Lee, 2012].

4.4. Melting Conditions Beneath BPVF

[28] Understanding the temperature and composition of the melting region and its relationship to the
continental lithosphere are keys to interpreting the processes that triggered melting below BPVF. Previous work used isotopic and trace element compositions of lavas from the Basin and Range to identify sub-continental lithospheric versus asthenospheric sources, and major element compositions to relate to depth of melting (i.e., tholeiitic shallow and alkalic deep [Ormerod et al., 1988; DePaolo and Daley, 2000]). Although this was a good initial approach, limited quantitative information was available on melting pressures (P) or temperatures (T).

[29] Given the primitive major element and water concentrations, $\text{Fe}^{2+}/\text{Fe}_T$ estimates, and assuming a peridotite source (based on section 4.3), we can estimate the melting P and T conditions beneath the BPVF using the thermobarometer developed by Lee et al. [2009]. The thermometer in this model is based on olivine-melt Fe-Mg exchange, while the barometer is based on Si activity in melts saturated with orthopyroxene+olivine. In practice, we start with major element compositions from BPVF whole rocks and melt inclusions that have passed the screening criteria discussed above (olivine-only crystallization; non-anomalous CaO; not pyroxenite source, etc.), add equilibrium olivine to them incrementally (taking into account $\text{Fe}^{3+}$; and using a constant $K_D = 0.3$ for the $\text{Fe}^{2+}/\text{Mg}$ exchange) until the melt is in equilibrium with mantle olivine (assumed to be Fo$_{90}$). For the H$_2$O content, we use the maximum concentration reported in melt inclusions from each vent. For samples <500 ka with no melt inclusion data, we used the average H$_2$O content (2 wt%) and $\text{Fe}^{3+}/\text{Fe}_T$ (0.25) of the nearby vents from which we have melt inclusion data. For samples >500 ky, which consist only of lavas unsuitable for melt inclusion volatile study (e.g., melt inclusions diffusively lose water in lava samples [Hauri, 2002; Portnyagin et al., 2008]), we assumed an average concentration of 1.5 wt% H$_2$O and $\text{Fe}^{3+}/\text{Fe}_T$ of 15%. We used those values taking in consideration upper mantle-like trace elements (see more discussions below). An uncertainty of 1 wt% H$_2$O introduces a temperature uncertainty of 20°C; and 10% $\text{Fe}^{3+}/\text{Fe}_T$ of 40°C and 0.3 GPa. Nevertheless, the T and P differences between samples <500 ka and samples >500 ky (below) cannot be explained by differences in H$_2$O and $\text{Fe}^{3+}/\text{Fe}_T$ (more details in Table S8 in the auxiliary material), but reflect distinct major element compositions (higher MgO, FeO and lower SiO$_2$) of the samples >500 ky compared to those <500 ky (Figure 4).

[30] The volcanics >500 ka include mostly the sequence of lavas exposed in Papoose Canyon [Blondes et al., 2008] and the Oak Creek lavas that bear mantle xenoliths [Beard and Glazner, 1995; Lee et al., 2000; Blondes et al., 2007]. These >500 ka samples form an array extending from ~1350°C and 2.2 GPa (~80 km) to 1280°C and 1.7 GPa (~60 km). Almost the entire variation is expressed in the Papoose sequence of lavas, with the earliest eruptives recording the highest P and T of mantle equilibration, and the latest ones recording the shallowest and coolest. Blondes et al. [2008] interpreted this to represent a single eruptive sequence, which could reflect efficient extraction of melt from over a 20 km depth range in the mantle. The calculated P and T derive from a clear increase in SiO$_2$ (from ~45–48 wt%) and decrease in FeO (from ~9.2–8.5 wt%) at nearly constant MgO (9.5–10.0 wt%) from stratigraphic bottom to top in the sequence (Figure 4). Moreover, the interdependence of T and P within the Lee et al. [2009] thermobarometer leads to error arrays that parallel the Big Pine data, and it is possible that some of these variations may be correlated errors related to olivine addition. For these reasons, we consider here the average melting conditions for these lavas (1320°C ± 20°C and 2.0 ± 0.2 GPa).

[31] BPVF volcanics erupted over the past 500 ka (which include all the scoria cones with melt inclusions reported here) record shallower and cooler melting conditions (1190–1250°C and 55–30 km). These conditions overlap, but are slightly shallower than the 68–46 km melting column estimated by Wang et al. [2002], in a melting inversion that did not take into account H$_2$O or $\text{Fe}^{3+}$. Different cones appear to record distinct P-T conditions (Jalopy cone shallowest and Goodale Bee cone deepest), and so there may be structure within the array, but again we will just consider the average here (1220 ± 20°C and 1.1 ± 0.2 GPa or 38 ± 7 km).

[32] Taken together, all of the Big Pine magmas equilibrated at temperatures lower than the dry solids of peridotite. This is consistent with the magmas containing 1.5–3.0 wt% H$_2$O, and so the melting process must be modeled with water. There are several recent parameterizations of wet melting, most of which follow a similar cryoscopic approach as outlined in Katz et al. [2003], Aubaud et al. [2004], and Langmuir et al. [2006]. Three parameters must be known: the cryoscopic effect ($\Delta T$ as a function of H$_2$O concentration in the melt, at constant F); the partition coefficient for H$_2$O between bulk mantle solid and melt (D$_{\text{H}_2\text{O}}$); and the melt productivity as a function of T (dT/dF). Two recent studies provide updated expressions for these parameters for peridotite + H$_2$O at a range of...
relevant pressures [Kelley et al., 2010; Hirschmann, 2010], and predict similar melting conditions, with temperatures within 20°C of each other (Table S8). At least in theory, given such a model for wet melting, along with independent constraints on the pressure and temperature of melting (from thermobarometer) and H2O (from melt inclusion measurements), it is possible to determine uniquely the melt fraction and water content of the source for a given primitive basaltic liquid. The >500 ka lavas record P and T of equilibration that is ~45 degrees below the dry peridotite solidus, in melts that we assume have ~1.5 wt% H2O, yielding 1–7% melting of mantle with 250–1150 ppm H2O in the source, respectively. This range of solutions is derived from both the Kelley et al. [2010] and Hirschmann [2010] models, given 20°C uncertainty in temperature. The <500 ka magmas reflect 8–10% melting and 1500–1900 ppm H2O in the source. Given the assumptions in H2O and Fe3+ in the older lavas, and in the mantle olivine composition (Fo90 for both), and the high temperature sensitivity of the modes, these ranges in F and source H2O may not be robustly different for the two groups of magmas. Nonetheless, the data set as a whole forms a strikingly coherent array that parallels melting isopleths (Figure 9). Whether this reflects some fundamental aspect of the melting process, or is a fortuitous outcome of the modeling and thermobarometry, is not entirely clear.

5. Discussion

5.1. The Water Content of the Mantle Beneath Big Pine

[33] The significant water concentrations in primitive BPVF melts (>1.5 wt% H2O) require a source of water in the mantle. The results of the melting model yield a wide range of results, from 250 ppm (at the upper end of MORB-OIB mantle, 100–300 ppm [Dixon et al., 2002]) to 1900 ppm H2O in the source. Given the assumptions in H2O and Fe3+ in the older lavas, and in the mantle olivine composition (Fo90 for both), and the high temperature sensitivity of the modes, these ranges in F and source H2O may not be robustly different for the two groups of magmas. Nonetheless, the data set as a whole forms a strikingly coherent array that parallels melting isopleths (Figure 9). Whether this reflects some fundamental aspect of the melting process, or is a fortuitous outcome of the modeling and thermobarometry, is not entirely clear.
We can arrive at a minimum Ce concentration in the BPVF source of 5.6 ppm by assuming depleted mantle Y concentration of 4.1 ppm [Salters and Stracke, 2004]. Given the restricted range in the H$_2$O/Ce ratio of Big Pine melt inclusions (164 ± 51), and the lack of fractionation between H$_2$O and Ce during crystallization or melting of anhydrous pyroxenes and olivines [Hauri et al., 2006], we calculate a minimum H$_2$O concentration in the Big Pine mantle of 925 ± 280 ppm. This minimum estimation falls near the mid-range of the permissible H$_2$O concentrations to satisfy the wet melting models above (i.e., 1075 ppm). Using the Ce source concentration above, and a D(Ce) of 0.015 (based on values in Kelemen et al. [2003]) and an average Ce concentration in Big Pine primary liquids of 97 ± 18 ppm for >500 ka volcanics and 78 ± 30 ppm for <500 ka volcanics (data from Blondes [2008] and this study, with olivine added until in equilibrium with Fo$_{90}$), the degree of melting is calculated to be 4.3 ± 0.9/–1.3% and 5.8 ± 4.5/–2.1%, respectively. This range in melt fraction (3.4–10.3%) not only is within the range of that calculated from the wet melting models (1–10 wt%), but also provides a better-defined mean around 4–6%. Thus, independent methods involving major, volatile and trace element data from lavas and melt inclusions, are consistent with average melting conditions beneath the BPVF involving mantle with ~1000 ppm H$_2$O melting to ~5% F. There are likely real differences in F and source H$_2$O with time, but given the uncertainties in the assumptions and models, we consider below only the implications of the average values.

[34] The above calculation provides a minimum H$_2$O content of the BPVF mantle (925+/−280 ppm) that rules out normal upper mantle as a source (100–300 ppm H$_2$O [Dixon et al., 2002]), and even exceeds the water storage capacity for nominally anhydrous minerals in the convecting upper mantle (<500 ppm H$_2$O [Hirschmann et al., 2009]). Thus, an additional source of water is required. There are several possibilities. One is crustal contamination. This seems unlikely at the BPVF, where melt inclusions with as much as 2 wt% H$_2$O occur in near primary Fo$_{90}$ olivines (e.g., the ones from Jalopy). There is no correlation between the H$_2$O content of melt inclusions, major elements, and Fo content of host olivines, as might be expected from a crustal assimilation process. Also, BPVF magmas bear mantle xenoliths [Ormerod et al., 1991; Beard and Glazner, 1995; Blondes et al., 2007; Kirby et al., 2008], and the lavas with clinopyroxene phenocrysts record mantle equilibration pressures [Mordick and Glazner, 2006]. Therefore, BPVF magmas had minimal residence in the crust. We thus find the crust an unlikely source of the excess H$_2$O. Mantle lithosphere may also be considered a source of water, in the form of hydrous minerals like amphibole and phlogopite, but it is noteworthy that none of the mantle xenoliths from Big Pine contain any hydrous phases [Beard and Glazner, 1995]. Moreover, our new water data for lithospheric mantle peridotite xenoliths demonstrate unusually dry mantle lithosphere beneath Big Pine, with bulk H$_2$O contents (<75 ppm; Figure 7) well below any of the above estimates (250–1900 ppm) for the melt source. Indeed, the mantle lithosphere beneath Big Pine appears to have been cold and dry, which may be why it has survived as an isotopically distinct layer since perhaps the Proterozoic [Beard and Glazner, 1995; Lee et al., 2000].

[35] We suggest that the excess water recorded in Big Pine magmas derives from the sub-lithospheric mantle. In fact, at the temperatures inferred for magma formation, mantle that contains significant water is weak and will flow as asthenosphere [Hirth and Kohlstedt, 1996; Karato and Jung, 1998; Karato, 2003]. There are three possibilities for the source of excess water in the mantle beneath Big Pine: hydrous minerals in the asthenosphere, subducting oceanic crust and the transition zone. Phlogopite is a Mg-rich mica and K richterite is a K-rich amphibole, both of which contain wt% concentrations H$_2$O in their structures and may be stable at pressures >4 GPa along a normal adiabat (Tp = 1350°C) in the upper mantle [Trønnes, 2002]. Phlogopite in the mantle beneath Big Pine may be responsible for the high time-integrated Rb/Sr reflected in the high $^{87}$Sr/$^{86}$Sr of the Big Pine magmas [Ormerod et al., 1991], and in the source of potassic volcanism that occurred in the Pliocene of the southern Sierras [Farmer et al., 2002]. On the other hand, both phlogopite and K richterite have a diagnostic H$_2$O/K$_2$O ratio of 0.38 (from stoichiometry) that can be used to test for their presence in the source of BPVF magmas. In a plot of H$_2$O/K$_2$O vs 1/K$_2$O [after Wallace and Anderson, 1998], BPVF melt inclusions appear to mix to phlogopite-K richterite for the more K-rich end-member, but they clearly mix to another, higher H$_2$O/K$_2$O component that is not part of the normal N-MORB-E-MORB array (Figure 10a). Arcs and back-arc basin magmas plot to very high H$_2$O/K$_2$O (>7) and may relate to the other end-member. Thus, while phlogopite-K richterite may exist in the source of BPVF magmas, another water source is still required.
As discussed above, the H$_2$O contents of Big Pine magmas approach those typical of back-arc basin basalts, where water is actively supplied from subduction zones [e.g., Kelley et al., 2006]. Indeed, BPVF melt inclusions plot in a distinct region on a H$_2$O/Ce-Pb/Ce diagram, with H$_2$O/Ce similar to upper mantle asthenosphere (MORB-OIB), but with clearly higher Pb/Ce, like volcanic arcs. Interestingly, continental arcs like the Cascades, Guatemala, and Mexico plot toward a mantle end-member with Pb/Ce higher than MORB-OIB, in the range of our new data from the BPVF (Figure 10b).

The trace element signatures of Big Pine magmas also share similarities with arc magmas (i.e., excess Pb, depletions in Nb and Ta; Figure 5). The dehydration of subducted material along slab P-T paths is an obvious supply of water to mantle above subducted slabs, but there is no subduction zone beneath Big Pine at this time. On the other hand, the shear wave tomography in Figures 11a and 11b illuminates the Isabella Anomaly, a prominent fast seismic anomaly located immediately to the west of the Sierra Nevada and BPVF. The Isabella Anomaly has been interpreted as actively foundering Sierran lithospheric mantle or lower crust [e.g., Zandt et al., 2004; Frassetto et al., 2011], or as a fragment of the Farallon Plate that did not detach [Pikser et al., 2012; Y. Wang et al., 2011]. This feature could drive mantle upwelling and melting, either as counterflow to the foundering drip, or induced flow as it is dragged northwest with the Pacific Plate. The Isabella Anomaly is one of the largest seismic anomalies in the western USA, comparable in magnitude to the actively subducting Gorda Plate to the north [Schmandt and Humphreys, 2010, 2011]. It is also possible this feature is still dehydrating, although subduction ceased at 20 Ma. In subduction zones, the K$_2$O/H$_2$O and H$_2$O/Ce of slab fluids is related to their last temperature of equilibration at the slab surface [Plank et al., 2009]. Interpreted in this way, the Isabella slab fluids would fall at the high temperature end (>950°C) of fluids that supply active subduction zones, consistent with the heating up of this slab remnant during long residence in the mantle.

Another possible source of water beneath Big Pine is from mantle that may upwell from the transition zone. The storage capacity of water in transition zone minerals is considerably higher (0.5–1 wt% [Ohtani, 2005]) than in the convecting upper mantle (<500 ppm H$_2$O [Hirschmann et al., 2009]) and so mantle that upwells from the

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**Figure 10.** Geochemical constraints on the source of water in BPVF magmas. (a) H$_2$O/K$_2$O versus 1/K$_2$O systematics [after Wallace and Anderson, 1998] suggest mixing between a deep volatile rich phase (Phlogopite/K Richterite) and an arc source for BPVF melt inclusions. (b) Melt inclusions from active volcanic arcs have uniformly higher Pb/Ce and H$_2$O/Ce than MORB-OIB, generally ascribed to a H$_2$O-Pb rich slab fluid or melt. Melt inclusions from the BPVF have high Pb/Ce like some arcs, but lower H$_2$O/Ce than most MORB-OIB. Some continental arcs (Cascades-Mexico) mix toward a mantle end-member with a high Pb/Ce like the BPVF. Blue circles are least degassed melt inclusions from BPVF (as in Figure 6). Gray circles include H$_2$O calculated from Cl and H$_2$O/Cl = 66 (as in Figure 6). Arc and back-arc data from Gribble et al. [1996]; Sadofsky et al. [2008]; Johnson et al. [2009]; Zimmer et al. [2010]; and Ruscitto et al. [2010].
Figure 11. Surface wave tomography beneath the Basin and Range and possible dynamic melting models for the BPVF. (a) Shear velocities at 70 km depth from Rayleigh-wave surface tomography [Rau and Forsyth, 2011]. Note the prominent Isabella Anomaly adjacent to BPVF. Melting contours and BPVF melt equilibration depths (red and blue bars) from Figure 9. LAB is the lithosphere-asthenosphere boundary deduced from the shear-velocity structure. Melting beneath the BPVF driven by (b) Extension-melt feedback, (c) small-scale convection related with lithospheric delamination and (d) the Isabella Anomaly as a slab remnant and small-scale convection related with it (see text for discussion). Mantle shear direction from Conrad et al. [2010].
transition zone will likely melt [Bercovici and Karato, 2003]. Assuming the maximum water storage, these melts will be water rich (>17 wt% $H_2O$ [Hirschmann et al., 2009]), and may ascend and supply water to the primary melting region beneath BPVF. Although there is no obvious dynamic reason why mantle might upwell from the transition zone beneath Big Pine, the high water contents in magmas throughout the Basin and Range [Plank et al., 2009] may require it, and mantle convection models may predict it as part of a large counterflow to Farallon downwelling [Moucha et al., 2008].

5.2. Implications of the Pressure and Temperature Estimates for Rheological Boundaries of Melt Equilibration

[39] Although mantle-melt thermobarometry provides valuable quantitative constraints on conditions within the upper mantle, the single pressure and temperature recorded in each magma composition is not always straightforward to connect to a melting process. One possibility is that magmas reflect only the final P and T of equilibration in the mantle. Melting prior to this point may have proceeded as a batch (melt remains with solid) or fractional (melt separates from solid) process, but if melt aggregates and equilibrates in one region, it will only reflect the final conditions. The other possibility is that the P-T recorded by magmas reflects a mean of the melts that have equilibrated at different depths. We find this latter scenario unlikely for the <500 ka BPVF magmas, given their very shallow pressures of equilibration (1.1 GPa or 38 ± 7 km, essentially at the Moho) and the lack of garnet signature in the magmas, meaning at least, that the contribution of deep melts to the mixture is minor.

[40] The pressures recorded in both series of melts may correspond to mechanical boundaries in the melting region. One is the lithosphere-asthenosphere boundary (LAB), which is unusually shallow, ~55–60 km as deduced from the maximum gradient in the shear velocity profile (Figure 11). Although the relatively high velocity (lithospheric) layer immediately beneath the Moho is too thin to be well resolved from the surface wave tomography alone [Rau and Forsyth, 2011], there is an apparent negative Ps conversion beneath Big Pine observed at ~55 km from receiver function analysis [Frassetto et al., 2011], also consistent with a shallow LAB. This boundary fundamentally separates the depths of equilibration of the two age groups of magmas. The older magmas (>500 ka) equilibrated at >65 km (70 km on average), possibly stalling at an older, slightly deeper LAB, at that time. The younger magmas (<500 ka) equilibrated uniformly above this boundary, in the inferred lithosphere, and on average at 38 ± 7 km, which is equivalent to crustal thickness here (of ~36 km) determined from receiver function analysis [Frassetto et al., 2011]. Thus, the older magmas equilibrated within asthenosphere, possibly at the base of an earlier ~70 km LAB, while the younger magmas have equilibrated above the modern ~55 km LAB, on average at depths close to the Moho. This is consistent with the fact that both the LAB and Moho are rheological or density boundaries, which may promote melt stalling and equilibration of melts that may have initiated at a much deeper solidus.

[41] It is important to consider that the younger magmas, which have equilibrated in the lithosphere near the Moho, clearly did not form there, as their temperatures are too high (1220°C) to sustain at the Moho without massive melting of the crust. Also, their temperatures exceed those recorded by the Oak Creek mantle xenoliths, 1000–1100°C at <1.5 GPa as reported by Lee et al. [2001] (although Ducea and Saleeby [1996] report up to 1200°C). Moreover, the lithosphere is too dry to be the source of these magmas, as recorded in the low water contents (<100 ppm $H_2O$) of the Oak Creek xenoliths (Figure 7), and the lack of hydrous minerals contained within them [Beard and Glazner, 1995]. Thus, the extent of re-equilibration within the lithosphere is partial, reflected in the pressures (which only require precipitation of olivine), but not the temperatures (which is retained in the transported melts) nor water contents (which would require equilibration with larger volumes of dry mantle). It is also important to keep in mind that the seismic results represent averages over significant areas; very localized thinning or erosion or infiltration of the lithosphere by melt would not be recognized.

5.3. Geochemical Indicators of Lithosphere Versus Asthenosphere Mantle Sources in the Evolution of BPVF

[42] Our thermobarometry results, in combination with the seismic structure of the upper mantle, point to different melt equilibration scenarios for the older and younger BPVF suites, with the >500 ka magmas equilibrating below the LAB in the asthenosphere, and the <500 ka magmas equilibrating within the lithosphere. This interpretation differs from previous work that considered all Big Pine
magmas to have a lithospheric source, based on their Proterozoic Sm-Nd model ages, their high $^{87}\text{Sr}/^{86}\text{Sr}$ (0.7054–0.7064), and their arc- or continent-like trace element signatures [Ormerod et al., 1988, 1991]. On the other hand, based on our new data here and that in Blondes et al. [2008], we find a systematic variation in diagnostic trace element ratios in both space and time. The hotter and deeper melts (>0.5 Ma) have higher Ce/Pb (14–21) and constant Ba/La values, closer to an upper mantle asthenosphere end-member (Figures 12a and 12b). Shallower and cooler melts (<0.5 Ma) have lower Ce/Pb (<14), and higher Ba/La (25–30) values, more typical of arc magmas and consistent with the available data from lithospheric mantle xenoliths from BPVF [Lee, 2005], suggesting a lithosphere component for these melts (Figures 12a and 12b). In the data set as a whole, there is a highly significant ($R^2 = 0.85$) correlation between Ce/Pb and the depth of equilibration, from 90 to 35 km. Thus there is a clear contribution from old, chemically distinct mantle that increases at shallower levels, consistent with a chemical boundary layer above the seismically imaged LAB (55–60 km depth, Figures 11b and 12a). On the other hand, the temperatures (>1200°C) and water contents (>1.5 wt% H$_2$O) of all the magmas seem inconsistent with formation in the cold and “dry” lithosphere.

Although the correlation is clear with trace elements (Ce/Pb, Ba/La), isotopic compositions do not vary systematically with depth, except within the Papoose section [Blondes et al., 2008]. The petrologic modeling together with the trace element correlations collectively suggest that both asthenosphere and lithosphere have played a role, and possibly evolved, over the formation of the BPVF magmas. It is possible that mantle interpreted as asthenosphere today (at 90 to 60 km depth from shear wave tomography in Figure 11b) was evolved from lithospheric mantle during melt infiltration and re-heating, thus explaining the old model ages of all BPVF magma sources [Beard and Glazner, 1995].

5.4. The Cause of Mantle Melting at Big Pine

Together, the petrological and seismological data provide a consistent view of the melting region beneath Big Pine. The base of the low velocity zone is around 225 km beneath Big Pine (Figure 11b),
which could correspond to the carbonatite-carbonated silicate melt transition [after Hirschmann, 2010]. This would occur at ~8 GPa (250 km) for mantle with 1000 ppm H2O and 250 ppm CO2 km and F = 0.1% (note that primary CO2 is generally not possible to constrain even from melt inclusions, which may already be saturated in a CO2-rich vapor when trapped; 250 ppm CO2 in the mantle is twice that needed to explain maximum CO2 contents in BPVF magmas, and so permissible and not unreasonable). The region of the lowest seismic velocities beneath Big Pine (<4.2 km/s) begins at around 120 km, which could correspond to the onset of significant silicate melting (F = 1% in mantle with 1000 ppm H2O at 4 GPa, 140 km). Melting then continues up to the base of the 500 ka LAB (~70 km), producing ~5% melt. Although the melting boundaries are tens of km deeper than the seismic boundaries, these differences are not likely significant given the uncertainties in both inversions. The >500 ka magmas then erupted rapidly above this depth, possibly by diking in cold lithosphere, carrying mantle xenoliths to the surface, with little residence in the crust. With time, the mantle lithosphere warms, magmas stall, react and equilibrate there. This leads to an erosion of the lithosphere, and possibly shallowing of the LAB from 70 to 55 km in ~500 ka. Crustal storage regions are likely near the brittle-ductile transition (~20 km), where magmas stall due to a rheological contrast, melt inclusions start to be trapped by crystal cooling (at ≤5 kb; Figure 6), magmas fractionate, and no longer bear mantle xenoliths. Volcanic vents form very near active faults in Owens Valley [Kirby et al., 2008], which enhanced ascent pathways.

[46] Despite this self-consistent view of the melting process, many questions still remain as to the ultimate cause of volcanism here. Why do mafic magmas erupt at Big Pine? What is special about the crust and mantle that leads to melting and eruption? There are many special tectonic features and events that appear to characterize the Big Pine region. (1) Active oblique strike-slip faulting and extension in Owens Valley [Phillips and Majkowski, 2011]; (2) proposed Pliocene foundering of adjacent Sierran lithosphere [Manley et al., 2000; Jones et al., 2004; Zandt et al., 2004]; (3) and the prominent Isabella anomaly to the west and thicker LAB to the east (Figure 11b). Which of these are the critical drivers for melting and eruption of mafic magmas here? And how can they be consistent with the constraints we have provided on the pressures, temperatures, water contents and evolution of the mantle melting region? We provide two views on melting, one top-down (1. Extension-Melt Feedback) and the other bottom-up (2. Small-Scale Convection).

[47] This view of the origin of the BPVF fundamentally relates melting to active lithospheric deformation. Deformation initiates upwelling that drives melting and starts a feedback process that also creates melt pathways to the surface. The broader implication here is that magmas in the Basin and Range will tend to form and erupt where deformation is active. This is true in a broad sense
(most of modern Basin and Range deformation and volcanism is occurring at its margins; along the eastern side of the Sierras (as in Big Pine) and along the western side of the Colorado Plateau [Bennett et al., 1998; Hammond and Thatcher, 2004]. Moreover, seismic images are providing abundant evidence for the presence of melt in the mantle over much of the Western USA, but in most places, it does not manifest at the surface (e.g., in the Amagmatic Zone of southern Nevada [Rau and Forsyth, 2011]), possibly because active deformation does not provide the melt enhancement, segregation and ascent pathways required for eruption. Further work linking melting in the mantle to surface deformation and volcanism across the Basin and Range will test these ideas. On the other hand, this view of deformation and melting does not provide a ready source of the excess water that is required in the source of Big Pine magmas. Normal Farallon subduction did occur in this region as recently as 18–20 Ma, and likely hydrated the mantle, but as we have argued above, the water concentration reflected in the source of Big Pine exceeds the normal storage capacity of anhydrous mantle at T_p = 1350°C (500 ppm), and phlogopite, which can be stable at these temperatures, cannot supply all the water (based on K2O/H2O, Figure 10).

2. Small-scale convection. The other driver of melting could be related to lithospheric drips and the Isabella anomaly to the west of the BPVF (Figures 11c and 11d). The Isabella anomaly is a fast seismic anomaly that extends to at least 200 km depth, and is located beneath the western Sierra foothills and Great Valley. Its origin is debated. Some ascribe it to Sierran lithosphere (garnet clinopyroxene lower crust and peridotite mantle) that foundered in the Pliocene [Zandt, 2003; Ducea and Saleeby, 1998; Jones et al., 2004] while others link it to the fossil Monterey microplate, a remnant of the Farallon Plate that was left when subduction ceased ~20 Ma (Wang et al., submitted manuscript, 2012). In the latter view, the Isabella anomaly is a slab fragment, still attached to the Pacific lithosphere, and possibly being dragged northwest along with it. In either case, a sinking or dragging structure (lithospheric drips or remnant subducting oceanic crust) may induce flow in surrounding mantle, either in an upward counterflow or in a small-scale convective circulation, in combination with the high mantle shear in the western Basin and Range [Zandt et al., 2004; Elkins-Tanton et al., 2001; Conrad et al., 2010]. It is also possible that the Isabella anomaly could supply excess water to the mantle melting beneath Big Pine, through heating-induced dehydration of hydrous Sierran lower crust or of hydrous Farallon slab. In this view, the thin LAB beneath Big Pine was not generated by extension, but by lithospheric foundering in the Pliocene, and melting occurs today due to an influx of water and convective upwelling. In this way, the BPVF is situated above a long-lived mantle wedge, like an arc, a southern extension of the ancestral Cascades (sensu [Cousens et al., 2008]). Although it is unclear whether Mesozoic arc crust or a slab subducted 20 my ago could retain enough water to supply the BPVF, the Isabella anomaly does appear spatially related to volcanism along its eastern margin, from Long Valley to Coso. On the other hand, this volcanism does not include the high-Mg# andesite compositions of baijites erupted in the south, where similar Guadalupe and Magdalena Farallon remnants have been proposed (Wang et al., submitted manuscript, 2012), nor the much lower 87Sr/86Sr of the ancestral or modern Cascades erupted in the north and related to Gorda and Juan de Fuca subduction [Cousens et al., 2008]. The isotopic composition of Big Pine magmas overlaps that of Sierra Nevada granites. The role of the Isabella anomaly as a water source remains to be tested by study of the water contents across the Basin and Range, to see how widespread or locally “wet” the mantle is.

6. Conclusions

Big Pine melt inclusions record surprisingly high H2O contents from 1.5 to ~3 wt.%, with a degassing/eruption path from >5 kb (~20 km) to the surface. Estimates of the oxidation state also indicate high Fe3+ contents (23–30% Fe3+/FeT), approaching the most oxidized arc magmas. On the other hand, lithospheric mantle xenoliths from the well-known Oak Creek location at Big Pine record water contents below 75 ppm, indicative of a dry lithosphere. Given a calculated mantle source of ~1000 ppm H2O for BPVF primary melts, the source of the water must be sub-lithospheric. Certain aspects of Big Pine magma chemistry resemble subduction zone magmas (e.g., Pb and Nb anomalies), while others do not (e.g., H2O/Ce; 87Sr/86Sr). One possibility is that the nearby prominent Isabella seismic anomaly is a Farallon slab remnant or lithospheric drip that supplied water to the region.

The BPVF samples also record a shallowing of mantle-melt equilibrium over the past 0.5 ka, from 2 GPa (65–70 km) and 1320°C to 1 GPa (35–40 km) and 1220°C. The depths of melting are also strongly
correlated with a shift over time to trace element ratios (Ce/Pb and Ba/La) diagnostic of subduction or lithospheric sources. Taken together, the magmas record asthenospheric melting deeper than 70 km during the early phase of volcanism, and shallower melt equilibration in a chemical boundary layer reflected in the more recent magmas. A strictly lithospheric source of the <0.5 Ma melts can be ruled out based on the high temperatures and water contents of the melts relative to the demonstrably lithospheric mantle xenoliths. Melt infiltration is thus a potentially important mechanism in affecting melt composition and in converting lithosphere to asthenosphere. The lack of a fast seismic lid in the mantle beneath Big Pine is consistent with the P-T-melting history recorded in the magmas, initiating as deep as 200 km in volatile-rich upwelling mantle melting history recorded in the magmas, initiating as deep as 200 km in volatile-rich upwelling mantle asthenosphere (T_p ~ 1350°C), infiltrating a 30 km thick chemical boundary layer (from 35 to 65 km), and equilibrating in some cases right up to the Moho. Such a scenario explains both the low velocity anomalies below the BPVF, and its geochemical and petrological evolution.

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